

[illegible]

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Use of shale oil sulfonic acids in the aniline dye industry.
A. M. Berkenheim and M. A. Berkenheim. *Org. Chem.*
Ind. IV, S. S. R. 7, 437-4(1940).—NH₂ sulfonates ob-
tained as by-products of shale-oil refining can be used in
place of soaps as emulsifiers. B. C. P. A.

Reaction between halogen derivatives of 6-methoxyquinoline and alcoholates. A. M. Berkenheim and L. V. Antik. *J. Gen. Chem.* (U. S. S. R.) 11, 637-40(1941). The purpose of the work was the prepn. of *N*-diethylaminoalkyl ethers of 7- and 8-hydroxy-6-methoxyquinolines. Acetylanisidine (180 g.) and 280 g. AcOH treated with 146 g. Br yielded a ppt. of bromoacetylanisidine, which was sapon. by boiling for 10 min. with 15% HCl to yield,

on neutralization with NH_4OH , 2-bromo-4-aminoanisole (I), m. 63-4° (in 44% yield). I (25 g.), 29 g. glycerol, 43.5 g. 96% H_2SO_4 and 20 g. Na arsenate heated, eventually to 160°, then treated with H_2O , filtered, neutralized and extd. with Et_2O , yielded crude 6-methoxybromoquinoline (II) which was purified through its HCl salt, yielding a mixt. of 6- and 7-Br compds., m. 78-103°, in 68% yield. By fractional soln. in Et_2O the less-sol. 8-methoxy-7-bromoquinoline (III) was obtained, m. 110-11° (8.5 g.). A soln. of 1.5 g. Na in 10.5 g. $\text{Et}_2\text{NCH}_2\text{CH}_2\text{OH}$ was treated with 10 g. III, heated to a final temp. of 180° through a period of 5.5 hrs., the melt dissolved in 5% HCl, neutralized, extd. by Et_2O and fractionated: the lower fraction, b.p. 113-14°, was identified as 6-methoxyquinoline, while some III was isolated from the higher fraction. 6-Methoxy-8-aminoquinoline (9 g.) in 30 g. 46.5% HBr treated with 29 cc. 15% NaNO_2 soln. at -10° fol-

lowed by 8 g. CuBr in 84 g. 48% HBr, heated to boiling for 15 min., treated with NaHCO_3 , yielded, upon purification through the HCl salt, 5 g. 6-methoxy-8-bromoquinoline (IV), m. 85-8°. A soln. of 5.5 g. Na in 33.5 g. $\text{Et}_2\text{NCH}_2\text{CH}_2\text{OH}$ treated with 45 g. 6-methoxy-8-bromoquinoline (V) and heated to 180° reacted spontaneously with a temp. rise to 212°; after 15 min. the product was worked up as above, yielding 13 g. 6-methoxyquinoline, V being obtained from the higher-boiling fractions. The same result was obtained on heating V with EtONa , as well as on heating IV with $\text{Et}_2\text{NCH}_2\text{CH}_2\text{ONa}$. It is concluded that heating of 7- and 8-halo-6-methoxyquinolines with alcoholates at 180° leads not to condensation but to replacement of nuclear halogen by H. The authors also prepd. 2-bromo-4-aminophenetole (analogously to I) and 6-methoxy-7-bromoquinoline (analogously to III), m. 80-90°. G. M. Kondapoff

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Condensation of 6-methoxy-8-hydroxyquinoline with 1,2-diethylamino-3-halopropanes. A. M. Berkenheim and N. S. Spasokukotskii. *J. Gen. Chem.* (U. S. S. R.) 11, 541-4 (1941).—In view of the usefulness of dialkylaminopolymethyleneamine derivs. of 6-methoxyquinoline as quinoline substitutes, the authors felt the need of prep. of simple dialkylaminopolymethylene ethers of 8-hydroxy-6-methoxyquinoline (I) in order to establish any antimalarial activity of 8-hydroxy-6-methoxyquinoline, which may result from hydrolysis of such ethers *in vivo*. Com. 6-methoxy-8-aminoquinoline (II) was purified through its HCl salt. Attempted conversion of II into I by diazotization and by Bucherer's method did not succeed. Attempts were made to prep. I by acid hydrolysis of II: 281 g. II in 360 cc. concd. H_2SO_4 and 880 cc. H_2O was boiled for 8 hrs. under reflux, filtered, and the sulfate thus obtained dissolved in H_2O and neutralized to give 210 g. of green crystals, m. 167-9° (from H_2O), which were found to be 6-hydroxy-8-aminoquinoline contaminated with 6,8-dihydroxyquinoline. Freshly distd. II (135 g.) in 190 cc. concd. H_2SO_4 and 160 cc. H_2O was refluxed (bath temp. 235-45°) for 34 hrs. (when a sample on diazotization and treatment with 2-naphthol gave a weak reaction); the mass was treated with H_2O , let stand, filtered, dissolved in H_2O and neutralized to yield 90% of 6,8-dihydroxyquinoline (III), as red crystals, m. 152-4°. Methylation of III yields a mixt. of isomers: 11 g. III in 40 cc. abs. EtOH is added to a soln. of EtONa (from 3.12 g. Na), then the mixt. is treated at 35° with 19 g. $p-MeC_6H_4SO_3Na$, which reacts with heat evolution; the $MeC_6H_4SO_3Na$ is filtered off and the filtrate acidified by AcOH and concd.; the residue, dissolved in 2% NaOH, washed with Et₂O and pptd. by AcOH, yields crude I, 124-5°. I (6.6 g.) in 80 cc. hot abs. EtOH is treated with 0.98 g. Na in 50 cc. EtOH, cooled, treated with 6.3 g. 3-diethylaminopropyl chloride, heated to 50° for 3-4 hrs., let stand, filtered and concd.; the residue is dissolved in Et₂O, washed with 1% KOH and distd. to yield 6-methoxy-8-diethylaminopropoxyquinoline, b.p. 198-205°, as a viscous liquid. The product was not active as an antimalarial.

G. M. Kozlovskii

ASH-514 METALLURGICAL LITERATURE CLASSIFICATION

RECORDS DIVISION

SEARCHED INDEXED

RECEIVED

NOV 1941

BERKENGEM, E. B.

BERKENGEM, E. B. "Investigation of the Tanning of Thin Gelatinous Layers with a 'Hydrotypical' Process." Min Culture USSR. Leningrad Inst of Cinema Engineers. Leningrad, 1956. (Dissertation for the Degree of Candidate in Technical Science)

So: Knizhnaya Letopis', No. 19, 1956.

BERKENGEM, B. M.

DECEASED

Chemistry

see ILC

BERKENGEYM, G.B.

Use of novocaine solutions in polyvinyl-pyrrolidone for prolonged
local anesthesia. Trudy 1-go MMI 3:175-184 '57. (MIRA 14:5)

(NOVOCAINE)

(VINYL COMPOUNDS)

(PYRROLIDINONE)

(LOCAL ANESTHESIA)

KOCHETKOVA, V.A.; BERKENGEM, G.B.; SIDEL'KOVSKAYA, F.P.

Prolongation of the effect of penicillin and streptomycin with the aid of aqueous solutions of polyvinylpyrrolidone. Antibiotiki 8 no.12:1100-1105 D '63.

(MIRA 17:10)

1. Gosudarstvennyy nauchno-issledovatel'skiy onkologicheskiy institut imeni Gertsena i Institut organicheskoy khimii imeni Zelinskogo AN SSSR.

C.A.		PROCEDURE AND REACTION		REMARKS	
BERKENGLEY, T. I.				2	
Application of the Fischer method for the determination of the solubility of water in organic compounds and in their mixtures at various temperatures. T. I. Berkengley, <i>Zvezdskaya Lab.</i> 10, 532-4 (1941). — The following method (C.A. 29, 6532) for the detn. of moisture was used with certain modifications to det. the soly. of water in various hydrocarbons, ketones, and their salts, and in other liquid org. substances at various temps. At temps. above zero the samples studied were satd. with water and kept in the thermostat for approx. 4 hrs., stirred constantly with water, and left in the thermostat at the same temp. until sepd. into layers. The upper layer (satd. with water) was decanted into another container, which was placed in the same thermostat. For temps. below zero the thermostat was filled with a mixt. of snow and water or with CaCl₂, and the thermostat vessel was placed in a large glass aquarium filled with approx. the same cooling mixt. In detn. of the soly. of water in ketones the alc. was replaced by pyridine as a solvent for 1 because ketones and aldehydes react with alc. The solubilities of water were: in aviation benzene, d₁₅ 0.8706, at -13, zero, 10, 20, and 40° 0.0075, 0.016, 0.021, 0.029, and 0.038 wt. %, resp.; in benzene produced by cracking, d₁₅ 0.8578, at -10, -1.5, 10, 20, and 40° 0.0085, 0.013, 0.020, 0.027, and 0.035 wt. %, resp.; in alkylbenzene, d₁₅ 0.8675, at -10, zero, 10, 20, and 40° 0.0063, 0.011, 0.014, 0.018, and 0.023 wt. %, resp.; in isooctane, d₁₅ 0.6947, at -1.5, 10, 20, and 40° 0.0011, 0.0037, 0.0055, and 0.0080 wt. %, resp.; in aviation gasoline B-70 at -5, 10, 20, and 40° 0.0011, 0.0033, 0.0045, and 0.0063 wt. %, resp.; in butyl alc., b. 116.5-117°, at -18, -7, 20, and 40° 2.80, 12.5, 23.1, and 20.4 wt. %, resp.; in MeCOEt (upper layer), d₁₅ 0.8112, at zero, 10, 20, 30, and 40° 22.5, 18.5, 17.2, 17.2, and 17.2 wt. %, resp.; in MeCOEt (lower layer) at zero, 14, and 40° 55.5, 53.5, and 74.4 wt. %, resp.; in MeCOEt, d₁₅ 0.8107, at zero, 20, and 40° 2.11, 8.59, and 3.02 wt. %, resp.; in the mixt. 80% MeCOEt + 20% MeCOEt, d₁₅ 0.8104, at -18, zero, 10, 20, and 40° 2.40, 4.95, 8.04, 6.20, and 6.20 wt. %, resp.; in isopropyl ether, d₁₅ 0.7394, at -20, -13, zero, 10, 20, and 40° 0.64, 0.70, 0.75, 0.80, 0.87, and 0.91 wt. %, resp.; in benzene, b. 80.4°, at 10, 20, 30, 40, and 50° 0.040, 0.053, 0.066, 0.084, and 0.114 wt. %, resp.; in cyclohexane, b. 80°, at 20 and 50° 0.0667 and 0.018 wt. %, resp. Although the results obtained for benzene and cyclohexane at above room temp. were somewhat lower than those obtained by Tarasenkova by the turbidity method (the method of Aleksey, modified by Rothmund), the caps. indicate that the Fischer method can be used for the detn. of water in a no. of substances not only at room temp. but also at higher temp. and at temps. below zero. Twelve references. W. K. Hean					
ASSOCIATE METALLURGICAL LITERATURE CLASSIFICATION					
EDUC. STUDY DIV.					
EDUC. DIV. CH. DIV. DIV.					
EDUC. DIV. CH. DIV. DIV.					

BERKENGEYM, T. I.

PA 196T15

USSR/Chemistry - Oxidants

Nov 51

"Theory of the Combined Action of Catalysts in Solution. III. Intermediate Products Under Catalysis of H_2O_2 With Salts of Calcium and Wolfram," G. A. Bogdanov, T. I. Berkengeym, Moscow

"Zhur Fiz Khim" Vol XXV, No 11, pp 1313-1317

Studied gasometric kinetics of catalytic decomposition of H_2O_2 in simultaneous presence of Ca salts and wolfrates. Established that process is case of ideal catalysis in which intermediate products form momentarily and reversibly.

196T15

USSR/Chemistry - Oxidants (Contd)

Nov 51

Kinetic curves indicated formation of 2 active intermediate products. Worked out methods for prep of intermediate products: yellow-green $CaNO_3$ and colorless $CaNO_6$.

196T15

PA 242T13

BERKENGEYM, T. I.

USSR/Chemistry - Hydrogen Peroxide Nov 52

"The Role of Intermediate Products in Catalysis by Means of a Deposit," G. A. Bogdanov and T. I. Berkengeym

"Zhur Fiz Khim" Vol 26, No 11, pp 1659-1663

Investigated the joint action of barium chloride and sodium tungstate in the form of a deposit on the decomn of H_2O_2 . Determined that the kinetic course of the catalytic reaction is very complex: at a relatively high initial concn of the substrate there was a zero-order reaction, then a quite abrupt decrease in the rate, a zero order reaction, and a slowing down of the rate at the end of the expt; at a relatively low initial concn of the substrate, there was a very abrupt decrease of the rate, a zero-order reaction, and a slowing down of the rate at the end of the reaction. This change in the order of the reaction brings with it also an unexpectedly abrupt decrease in the vol and a change in the color of the deposit. The authors surmise that the observed effects are caused by the formation of certain intermediate compds of different properties and compn. As a result of this work, they isolated hitherto unknown barium pertungstates, established their compn, and developed a method for obtaining them.

242T13

B 9

-Berkengeym, T.I.
Category: USSR

Abs Jour: Zh--Kh, No 3, 1957, 7526

Author: Bogdanov, G. A., Berkengeym, T. I., and Sherbinin, V. A.

Inst : Not given

Title : Additional Materials on the Theory of the Joint Action of Catalysts in Solution. I. Intermediate Products of the Decomposition of H_2O_2 Catalyzed by Calcium and Molybdenum Salts

Orig Pub: Zh. Fiz. Khimii, 1956, Vol 30, No 4, 889-895

Abstract: The gasometric method (G. A. Bogdanov, Zh. fiz. khimii, 1950, Vol 24, 1450; 1951, Vol 25, 323) has been applied to the investigation of the homogeneous catalytic decomposition of H_2O_2 using a mixture of $CaCl_2$ and Na_2MoO_4 . The rate of decomposition vs. H_2O_2 concentration curve passes through a maximum independently of the temperature and the H ion concentration. The shape of the

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Category: USSR

B-9

Abs Jour: Zh--Kh, No 3, 1957, 7526

kinetic curves remains unchanged in the absence of CaCl_2 but a sharp increase is observed in the reaction rates. The authors explain the relationships observed by the formation of two intermediate substances of varying peroxide oxygen content. Two salts with the following compositions have been isolated from the reaction mixture: $\text{Ca}_2\text{Mo}_2\text{O}_{13} \cdot 9\text{H}_2\text{O}$ and $\text{CaMoO}_8 \cdot n\text{H}_2\text{O}$.

Card : 2/2

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Additional material in the file
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without prior approval of the
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The material is classified "Secret"
because it contains information
the disclosure of which could be
detrimental to the national
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APPROVED FOR RELEASE: 06/08/2000

CIA-RDP86-00513R000204920012-1"

BERKENEYM-SHCHEKINA, A.S.

Organizational aspects of complete deratization in Moscow; summary of a report. Trudy probl. i tem.sov. no.5:88-92 '55. (MIRA 8:12)

1. T'Sentral'naya kontrol'no-issledovatel'skaya laboratoriya Mosgor-dezstantsii

(Moscow--Rodent control)

BERKES, B.

"Galvanometers. II" (To be contd.) p. 282
(ELEKTROTEHNIŠKI VEŠTNIK, Vol. 21, no. 9/10, 1953, Ljubljana, Yugoslavia)

SO: Monthly List of East European Accessions, LC, Vol. 3, no. 5, May 1954/Uncl.

BURKES, R.

"Galvanometers. III." p. 327

(ELEKTROTEHNIŠKI VESTNIK, Vol. 21, no. 11/12, 1983, Ljubljana, Yugoslavia)

SO: Monthly List of East European Accessions, LC, Vol. 3, no. 5, May 1954/Uncl.

BERKES. B.

Computation of intermediate-frequency filters.

p. 237
Vol. 23, no. 7/8, 1955
ELEKTROTEHNIŠKI VESTNIK
Ljubljana

So; East European Accessions List (EEAL), LC. Vol. 5, no. 2, Feb. 1956

BERKES, B.

Parallel oscillating circuit. I. (To be contd.) p. 24. ELEKTROTEHNSKI
VESTNIK. (Institut za elektrisko gospodarstvo, Fakultata za elektrotehniko
in Institut za elektrozeve) Ljubljana. Vol. 24, no. 1/3, Jan./Mar, 1956.

So. East European Accessions List Vol. 5, No. 9 September, 1956

BERKES, B.

Optimal number of windings of a loaded coil with induction current. In German. p. 191.

GLASNIK MATEMATICKO-FIZICKI I ASTRONOMSKI. PERIODICUM MATEMATICO-PHYSICUM ET ASTRONOMICUM. (Društvo matematičara i fizičara Hrvatske i Prirodoslovno-matematički fakultet Sveučilišta u Zagrebu) Zagreb, Yugoslavia. Vol. 13, no. 3, 1958.

Monthly List of East European Accessions (EEAI), LC, Vol. 9, no. 2, 1960. Uncl.

BERKES, Branko, Ing. (Zagreb)

Direct-current voltage stabilizers with a high stability factor.
Elektr vest 27 no.9/10:316-318 S-0 '59. (EEAI 9:10)

1. Institut "Ruder Boskovic", Zagreb
(Voltage stabilizers)

YUGOSLAVIA

Lj. POLJAKOVIC, J. ROSIC and L. BERKES. Department of Biochemistry (Institut za Biohemiju) School of Pharmacy (Farmaceutski Fakultet) Belgrade.

"Determination of Alkali-Resistant Hemoglobin."

Belgrade, Arhiv za Farmaciju, Vol 12, No 5, 1962; pp 309-312.

Abstract [English summary modified]: Hemolysates of erythrocytes of unspecified number of healthy adults were 100% resistant to alkaline denaturation; all of 12 fetal (umbilical cord blood) hemolysates were susceptible to 50 to 70%. Mixing adult and fetal bloods gave results as anticipated except in the 1:4 ratio which gave high error rate. Method is considered reliable and fast enough for clinical use. Four tables, 5 references: 1 German thesis, 4 English.

1/1

BERKES, I.; DEBETTER, I.; KOSTKA, I.

"The 1 mv. Van De Graaff accelerator of the Section of Nuclear Physics."
p. 209.

MAGYAR FIZIKAI FOLYOIRAT. (Magyar Tudományos Akademia). Budapest, Hungary,
Vol. 6, No. 3, 1958.

Monthly list of East European Accessions (EEAI), LC, Vol. 8, No. 8,
August 1959.
Uncla.

HUNGARY/Nuclear Physics - Installations and Instruments. Methods C-2
of Measurement and Research

Abs Jour : Ref Zhur - Fizika, No 4, 1959, No 7404

Author : Barkes Istvan, Demeter Istvan, Kostka Pal

Inst : Atomfizikai Osztaly, Hungary

Title : Van de Graaf Accelerator for 1 Mv, of the Division of the
Atomic Physics of the Central Research Physics Institute

Orig Pub : Magyar fiz. folyoirat, 1956, 6, No 3, 209-223

Abstract : Description of the power supply and the accelerating portion
of the Van de Graaf 1 Mv generator. The generator accelerates
electrons to energies of 200-800 kev. The maximum current
of the accelerated particles reaches 40 ma. The voltage is
measured with the aid of a deflecting magnet with an accur-
acy of + 1.2 percent. The maximum current of the deflected
beam is 8 ma. It is possible to operate with the beam both
in vacuum and in air.

Card : 1/1

BERKES, I.; DEMETER, I.; KOSTKA, P.

"The IMV Van de Graaf generator for the acceleration of electrons."

p. 268 (Energia Es Atomtechnika) Vol. 10, no. 5/6, Aug. 1957
Budapest, Hungary

SO: Monthly Index of East European Accessions (EEAI) LC. Vol. 7, no. 4,
April 1958

HUNGARY/Nuclear Physics - Installations and Instruments.
Methods of Measurement and Research.

C

Abs Jour : Ref Zhur Fizika, No 12, 1959, 26655

Author : Berkes, I.

Inst : Central Research Institute of Physics, Hungarian
Academy of Sciences, Budapest, Hungary

Title : Effect of Magnetic Stray Field on the Location of
Image in Nuclear Spectrometers.

Orig Pub : Acta phys. Acad. scient. hung., 1958, 9, No 1-2, 13-
21

Abstract : In transverse magnetic spectrometers of the sector
type, the position of the image does not coincide
with the calculated value, owing to the action of
the stray field. The shift of the image in the trans-
verse direction leads to difficulties in determination

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HUNGARY/Nuclear Physics - Installations and Instruments.
Methods of Measurement and Research.

C

Abs Jour : Ref Zhur Fizika, No 12, 1959, 26655

of the momentum of the focuses particles from the value of the magnetic induction and geometric parameters of the spectrometer. In practice this shift is compensated for by a certain change in the intensity of the magnetic field. The author calculates the transverse shift and gives formulas for the value of the change in the field, necessary for its compensation. The results of the calculations were verified against measurements of the conversion line Cs137 on the transverse magnetic spectrometer with an angle of rotation of the electron beam by 90° , and was found to be in good agreement with the experiment. -- K.P. Mitrofanov

Card 2/2

- 7 -

BERKES, Istvan; DEMETER, Istvan

Small angle scattering of electrons. Koz fiz kozl MTA 7 no.4:203-217
'59. (EEAI 9:8)

1. A Magyar Tudományos Akademia Kozponti Fizikai Kutato Intezete,
Atomfizikai Osztaly.
(Electrons)

BERKES, Istvan; SZABO, Ferenc

The 2d Geneva Atomic Energy Conference. Fiz szemle 9 no.2:39-43 F '59.

1. Kozponti Fizikai Kutato Intezet.

BERKES, I.

"The exhibition of nuclear power in Geneva." p. 175.

ENERGIA ES ATOMTECHNIKA. (Energiagazdalkodasi Tudomanyos Egyesulet).
Budapest, Hungary, Vol. 12, No. 2/3, Feb./Mar. 1959.

Monthly list of East European Accessions (EEAI), LC, Vol. 8, No. 8,
August 1959.
Uncla.

~~BERKE~~ Istvan; DEMETER, Istvan; FODOR, Ilona; KESZTHELYI, Lajos

Voltage calibration of the K-800 cascade generator. Koz fiz kozl MTA
8 no.2/3:167-171 '60. (EEAI 10:4)

1. A Magyar Tudományos Akademia Kozponti Fizikai Kutato Intezete,
Magfizikai Laboratorium I.
(Cascades)

BERKES, Istvan; KESZTHELYI, Lajos

Measuring the durability of atomic nuclei. Fiz szemle 10 no.9:262-268 S '60.

1. Kozponti Fizikai Kutato Intezet, Budapest. 2. "Fizikai Szemle" szerkeszto bizottsagi tagja (for Keszthelyi).

BERKES, Istvan; DEMETER, Istvan; DEZSI, Istvan; L. FODOR, Ilona; KESZTHELYI,
Lajos

Investigations in the field of the background reduction of scintillation
counters. Koz fiz kozl MTA 9 no.3:165-169 '61.

1. Magfizikai Laboratorium I.

26752
G/016/61/009/004/001/001
B120/B108

21.6000

AUTHORS: Keszthelyi, L., Berkes, I., Demeter, I., and Fodor, I.
(Budapest)

TITLE: Absolute measurement of the number of γ -quanta

PERIODICAL: Experimentelle Technik der Physik, v. 9, no. 4, 1961,
197 - 203

TEXT: The authors applied the γ - γ coincidence method to scintillation counters with electronic rate meter in order to determine their absolute efficiency for 4.43 and 11.67 Mev γ -radiation produced in a $B^{11}(p, \gamma)C^{12}$ reaction. If the absolute efficiency of the detector is known, it is possible to determine the absolute number of γ -quanta emitted per unit time from the number recorded by the detector. This is important for the determination of various capture cross sections and for the evaluation of the cross section of photonuclear reactions. For the measurement, a 1 cm thick B target was placed between two scintillation counters with NaI(Tl) crystal (3 by 3" and 1.5 by 1.5", respectively).

Card 1/5

Absolute measurement of the ...

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B120/B108

The front side of the one counter was 6 cm, that of the other 4.5 to 24 cm distant from the center of the target. The target was bombarded by 170-kev protons perpendicular to the connecting line of the two counters. Fig. 1 shows the block diagram of the electronic measuring arrangement. The discriminator I was used as an integrating stage. Pulses with amplitudes exceeding 7 Mev were detected in the measurement of the 12-Mev group. The discriminator II was adjusted to the maximum of the amplitude distribution of 4.43 Mev. The delay circuit was empirically adjusted to the maximum coincidence number. The signals from the individual counters were led to a 100-channel analyzer (A I 100-1) with ferrite storage system. At first, two measurements were made in which the counter with the large crystal was 4.5 and 24 cm from the target. The results had to be corrected, since about + 10% of the 11.7-Mev radiation was also covered in counter II, and some 3% of the 16.1-Mev radiation in counter I. Random coincidences had to be considered, too. The corrected results read as follows (n in pulses/sec, K = coincidence). 4.5 cm distance : $n_I = 2550 \pm 8$, $n_{II} = 338 \pm 18$, $n_K = 7.37 \pm 0.15$. 24 cm distance : $n_I = 244 \pm 11$,

Card 2/5

Absolute measurement of the ...

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G/016/61/009/004/001/001
B120/B108

$n_{II} = 305 \pm 16$, $n_K = 0.67 \pm 0.027$. According to the anisotropy of the angular correlation of the observed γ -radiation of the 4-Mev and the 12-Mev group, respectively, one finds: $n_I = \omega_I \eta_{I,12} n_0 w(\theta_1)$; $n_{II} = \omega_{II} \eta_{II,4} n_0 w(\theta_2)$; $n_K = \omega_I \omega_{II} \eta_{I,12} \eta_{II,4} w(\theta_1, \theta_2, \phi)$. (ω = solid angle of the counter, referred to the radiation source and divided by 4π ; η = efficiency; n_0 = number of cascade processes/sec; θ_1, θ_2 = polar angles of the two radiations (both 90°); ϕ = angle formed in the horizontal plane by these radiations (180°); $w(\theta_1)$, $w(\theta_2)$, and $w(90^\circ, 90^\circ, \phi)$ were taken from publications. The authors obtained the following results. Distance 4.5 cm: $\omega_I \eta_{I,12} = (1.98 \pm 0.11) \cdot 10^{-2}$, $\omega_{II} \eta_{II,4} = (2.71 \pm 0.13) \cdot 10^{-3}$. Distance 24 cm: $\omega_I \eta_{I,12} = (1.98 \pm 0.12) \cdot 10^{-3}$, $\omega_{II} \eta_{II,4} = (2.55 \pm 0.13) \cdot 10^{-3}$. The following also holds: $\omega \eta = \frac{1}{2} \int_0^{\pi/2} (1 - e^{-\mu \sin \alpha}) \sin \alpha d\alpha$. μ denotes the

Card 3/5

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B120/B108

Absolute measurement of the ...

absorption coefficient of the NaI crystal for a radiation of given energy,
 $\alpha_2 = \arctan \frac{r}{t}$, r = radius of the NaI crystal, t = distance between its
 front plate and the radiation source, $K(\alpha) = t \sec \alpha$ for $0 \leq \alpha \leq \alpha_1 = \arctan \frac{r}{h+t}$,
 $K(\alpha) = r \operatorname{cosec} \alpha - h \sec \alpha$ for $\alpha_1 < \alpha < \alpha_2$, h = height of the
 NaI crystal. Fig. 4 shows the curve for 11.7-Mev γ -quanta, obtained by
 graphical integration of this equation, which agrees within $\pm 10\%$ with the
 values obtained at various distances of the counter with the large crystal
 from the radiation source. There are 4 figures, 2 tables, and 9 refer-
 ences: 1 Soviet-bloc and 8 non-Soviet-bloc. The four most recent refer-
 ences to English-language publications read as follows: Ref.2: T. Naka-
 mura, K. Futumaga, K. Takamatsu and S. Yasumi, Mem. Coll. Sci. Kyoto Univ.
 29, 141 (1959); Ref.4: J. L. Putman, Measurement of Disintegration Rate, in
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 Lauritsen, Nucl. Phys. 11, 1(1959); Ref.9: P. J. Grant, F. C. Flack, J. G.
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Absolute measurement of the...

26752
G/016/61/009/004/001/001
B120/B108

ASSOCIATION: Zentralforschungsinstitut für Physik, Budapest (Central
Research Institute of Physics, Budapest)

SUBMITTED: January 24, 1961

Legend to Fig.1: (1) amplifier; (2) discriminator; (3) coincidence; (4) de-
lay circuit; (5) counter; (6) multichannel analyzer; (7) amplifier; (8) de-
lay line.

Card 5/5

BERKES, Istvan; KESZTHELYI, Lajos

Measuring the durability of atomic nuclei. Fiz szemle 10 no.9:
262-268 8 '61.

1. Kozponti Fizikai Kutato Intezet, Budapest 2. "Fizikai Szemle"
szerkeszto bizottsagi tagja (for Keszthelyi).

35398

H/016/62/000/003/001/003
D249/D301

24,6600 (1057,1482)

AUTHOR: Berkes, István

TITLE: Investigating the structure of atomic nuclei by the scattering of high energy electrons

PERIODICAL: Fizikai szemle, no. 3, 1962, 69-74

TEXT: Research concerning atomic nuclei was based only on the study of nuclear interactions till 1950. These investigations resulted in a well-known relation between the radius and mass, suggesting that protons and neutrons are uniformly mixed in the nuclei. R. Hofstadter and co-workers pioneered the study of the structure of nuclei by investigating the scattering of high energy electron beams. The experimental technique and results of Hofstadter are reviewed. At first the theory of electron scattering is discussed. According to the considerations of classical mechanics the angle of scattering is proportional to the distance between the electron and the nucleus. If the distance is large the field of the electron cloud surrounding the nucleus screens its field and the incident beam

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Investigating the structure of ...

H/016/62/000/003/001/003
D249/D301

of electrons is scattered only by a weaker field. The electrons scattered at small angles indicate the change of potential along their path and the distribution of density of the electron cloud. If the incident electron beam passes close to the nucleus, the potential inside the nucleus does not tend to infinity as in an ideal Coulomb field but decreases. Consequently the angle of scattering decreases which was noted in the early experiments. In a quantum mechanical treatment the differential cross section of the effect can be computed from Dirac's equation. Applying the Born approximation, an explicit relation can be obtained between the differential cross section of the elastic scattering and the potential causing the scattering or the charge distribution originating the potential. Two shortcomings of the application of the Born approximation are pointed out: 1) No scattering occurs at certain angles; 2) the Born approximation attributes the same wavelength to the electrons inside the nucleus and outside. In heavy atoms the electronic wavelength decreases inside the nucleus. In case of energies of several hundred MeV and heavier nuclei the radii computed by this approximation are larger than in reality. Consequently numerical techniques are used for computations. Results of the measurements for the charge distribution in nuclei are discussed. As uniform basis of comparison the 'square mean radius' is defined as

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Investigating the structure of ...

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D249/D301

Eq. 5 $\bar{r}_{nk} = \int_0^{\infty} r^2 \rho(r) 4\pi r^2 dr$, r being the radius, ρ - distribution of charge. Dimensionless coordinate $y = r/r_{nk}$ is introduced and the $p(y)$

charge distributions are discussed. Analytical expressions are given for the following charge distributions: 1) Square, 2) Gaussian, 3) modified Gaussian, 4) exponential, 5) modified exponential, 6) Fermi distribution. With the application of these charge distributions $\sigma(\theta)$ can be computed where θ is the angle of scattering. This can be compared with the results of the measurements. In evaluating high energy scattering experiments the charge distribution cannot be obtained by direct computation from the differential cross-section. The numerical results of Hofstadter are reviewed and tabulated. The differential cross-section is computed for scattering by protons. Finally the inelastic scattering of electron beams is reviewed. The application of a magnetic analyzer made it possible to study the energy spectra of the scattered electrons. If inelasticity scattered electrons cause nucleon emission, they will have a somewhat diffuse energy spectrum. The width of the latter depends on the distribution of impulse of the nucleons inside the nucleus. The theoretical

Card 3/4

Investigating the structure of ...

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D249/D301

interpretation of the investigations reviewed above is based on the theory of fields. The problem could arise that in case of very small distances (approximately 10^{-13} cm) the laws of quantum electrodynamics are not applicable any more. There are 7 figures, 2 tables and 5 references: 1 Soviet-bloc and 4 non-Soviet-bloc. The 4 most recent references to the English-language publications read as follows: Lyman, Hansen, Scott, Phys. Rev. 84,626 (1951); R. Hofstadter, Rev. Mod. Phys. 28,214 (1956); R. Hofstadter, F. Bumiller, MR. Yearian, Rev.Mod.Phys. 30,482 (1958); I. Berkes, I. Demeter, Nucl.Phys., 15, 421, (1960).

ASSOCIATION: Közponi fizikai kutató intézet (Central Institute of Research in Physics)

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BERKES, Istvan

Structural investigation of atomic nuclei by scattering high-power electrons. Term tud kozl 6 no.8:346-349 Ag '62.

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BERKES, Istvan

Election irradiation from the point of view of physics. Energia es
atom 15 no.6:254-258, 280 Je '62.

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BERKES, István, tudományos munkatárs; FODOR, Ilona, tudományos
munkatárs, Koz fiz közl MTA 12 no.4:299-312 '64.

1. Central Research Institute of Physics, Hungarian Academy
of Sciences, Budapest.

BERKES, IVAN

③

Electrophoresis on paper. 1. Influence of the conductivity of the buffer on the degree of separation of the components of a mixture of proteins. Ivan Berkes and Vinka Karys (Med. Fac., Zagreb, Yugoslavia). *Arch. Kém.* 24, 73-82 (1952).—The influence of the cond. of the buffer on the mobility of colloid ions on filter paper (Whatman No. 1) has been studied for 4 buffers: Michaelis buffer, veronal-veronal-Na, Sørensen phosphate buffer, and a borate buffer of 12.4 g. H_3BO_3 , 56 ml. N NaOH, and 160 ml. of NaCl. The sepn. of human serum albumin from γ -globulin is used as the basic measure of separability (10 hrs.). For comparison of the buffers this new concept, the separability U_n , is defined as $U_n = D/H \cdot h$, where D = distance travelled by the colloid, H = potential drop across the paper in v./cm., and h = time in hrs. The expts. were done at 23°, the paper was 4×45 cm. U_n rises if the cond. decreases but shows no obvious relation to the ionic strength. Thus, the suitability of buffers is not detd. by their ionic strength; both phosphate and borate buffers can be used in paper electrophoresis, if the conductivities are adjusted to a corresponding value. If an elec. current passes through a filter paper, the following will occur simultaneously: H_2O evaps. from the paper, the cathode and anode buffer vessel will be emptied by capillary syphoning, the electrophoresis of the colloid takes place, there is an ionophoresis of the cations and anions, and the cond. and current along the paper changes with the potential gradient which gives rise to the other phenomena.

Werner Jacobson

BERKES, I.

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BERKES, I.

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V. Mihajlov

BERKES, Ivan

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May 25, 1954
Biological Chemistry

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Fibrinogen determinations by paper electrophoresis.
Ivan Berkes and Vinka Karas (Univ. Zagreb, Yugoslavia).
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independent component in the paper electrophoresis of
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be established: the analogous and the selective types. In
the former the distribution of protein fractions is analogous
to that of the plasma electrophoretic pattern. The selec-
tive type shows a specific permeability of cell walls to some
protein fractions especially to albumin, α - and γ -globulins.
The Rivalta reaction was also investigated. The ppt. does
not contain any proteins but consists of equimolar parts of
glucosamine and glucuronic acid. 27 references.
V. Mihailov

BERKES, I

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MD

②

Quantitative assay. I. Berkov, R. Berkes-Tomasovik.

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DUKIC-TADIC, Mirjana; BERKES, Ivan; POLJAKOVIC, Ljubica; PETRONIJEVIC,
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Histamine and pregnandiol excretion in normal and complicated pregnancy with special reference to puerperium. Srpski arh. celok. lek. 92 no.12:1183-1189 D '64.

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DANILOV, V.P., red.izd-va; ASTAF'YEVA, G.A., tekhn.red.

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1. Institute of Medical Physics, Budapest.

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for photoconductivity. Magy fiz folyoir 8 no.6:519-524 '60.
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1. Orvosi Fizikai Intezet, Budapest.
(Salt) (X rays) (Photoconductivity)

BERKES, Laszlo, okl.mernok

Short history of the reconstruction of the Danube Bridge at
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15-17 Ja '63.

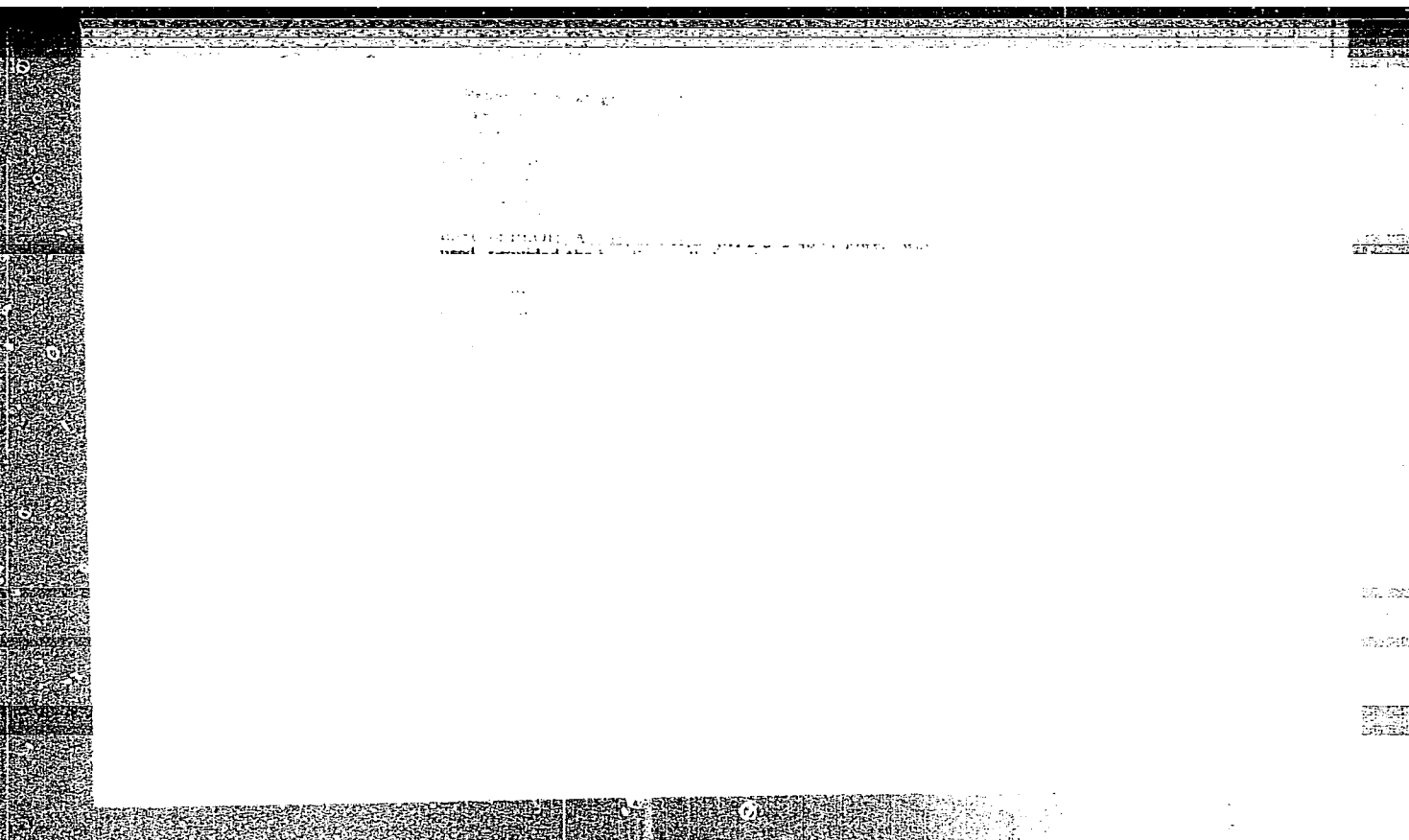
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(BLOOD PROTEINS, determ.

electrophoresis, filter paper technic. (Ser))



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BERKES, S.

(2)
Examination of the content of a cyst. M. Eller-Herman and S. Berkes. *Acta Pharm. Jugoslav.* 3, 119-20(1953) (German summary).—The content of an ovarian and peritoneal pseudomyxoma was examd. for its chem. compn. and homogeneity. The material contained 9% of dry matter, 1/3 of which could be pptd. with AcOH. The ppt. possessed characteristics of a mucoprotein but could not be classified as a pseudomucin according to Hammarsten. The ppt. existed in the cyst in the form of an acid salt. Electrophoresis showed that the mucoprotein was not homogeneous but contained at least 2 components. The nonpptd. part possessed the characteristics of a carbohydrate. 10 references. V. Mihajlov

BERKES-TOMASEVIC, P.; MATVEJEVA, L.

Biochemical changes in alcoholism. Acta med. iugosl. 15 no.2:
208-213 '61.

(ALCOHOLISM blood)

YUGOSLAVIA

P. BERKES-TOMASEVIC, J. ROSIC and M. IGNJATOVIC, Department of Chemistry of Veterinary Faculty (Hemijski institut Veterinarskog fakulteta) and Department of Biochemistry of the Faculty of Pharmacy (Biohemijski Institut Farmaceutskog fakulteta), Belgrade.

"Quantitative Starch Gel Electrophoresis."

Belgrade, Arhiv za Farmaciju, Vol 13, No 1, 1963; pp 9-17.

Abstract [English summary modified]: Description of method using domestic starch requiring 3 - 4 hours' hydrolysis; thin layer permits both quantitation by densitometry as for paper electrophoresis and storage for record; 100 pathologic human sera were so studied with excellent results. Diagram of apparatus, 6 electrophoresis patterns and 4 curves; 20 Western references.

1/1

YUGOSLAVIA

Persida BERKES-TOMASEVIC and Vera TERZIC, Department of Chemistry of Veterinary Faculty of University (Hemijski institut Veterinarskog fakulteta Univerziteta) Head Docent Dr Eng Persida BERKES-TOMASEVIC, Belgrade.

"Standardization of Hemoglobinometry."

Belgrade, Srpski Arhiv za Celokupno Lekarstvo, Vol 91, No 2, Feb 63; pp 149-156.

Abstract [English summary modified]: Authors worked up the cyanmethemoglobin method and determined its suitability on 52 blood specimens. Precise technical details of method are given as recommended for standard adoption in Yugoslavia. Formulae, table, graph; 35 Western references.

1/1

BERKES, Z.

3

(2) Geo

Meteorological Abst.
Vol. 4 No. 10
Oct. 1953
Part 2
Bibliography on
Auroras

4J-120

Berkas, Zoltan and Florian, Endre, Sarkifeny
Magyarorszagon 1938, januar 25-26-an. [Aurora in
Hungary on Jan. 25-26, 1938.] Idojaras, 42 (5/6):
117-123, May 1938. 4 figs., 4 refs. German summary
p. 145. DLC--This contribution to the rich observa-
tional material available for the intense auroral
activity of Jan. 25-26, 1938 comes from a country
where auroras are rarely seen. In a summary, based on
some 200 individual eye-witness reports, drawings and
photographs, the general development of the phenomenon
is described. From 8:14 p.m. Jan. 25 to 3:30 a.m. Jan.
26 a total of 7 displays were observed, two of them
with a luminosity unprecedented in Hungary since 1870.
The location of the aurora is put between 50 and 55°N
lat at a height of 200-300 km. Accompanying magnetic
disturbances, radio propagation anomalies and static
are mentioned. Earlier observations of auroras in
Hungary are listed. Auroral theory is reviewed briefly.
Subject Headings: 1. Aurora of Jan. 25-26, 1938
2. Low latitude auroras 3. Hungary.--G.T.

551.594.5(439.1)

[illegible]

Berkes, Zoltan

3.6-272
 Berkes, Zoltan. A napfoltok és az időjárás (Sunspots and weather). Természettudományi Közlemények 111(7):443-445, July 1952. DLC. Calculation of the sunspot number for the spring of 1952 is interpreted as a function of the sunspot number. The paper discusses the relationship between the sunspot number and the weather. It is noted that the sunspot number is a measure of the solar activity, and that the weather is a function of the solar activity. The paper concludes that the sunspot number is a good indicator of the weather. *WC*

18-year recurrence of phases of the moon is also observed. The importance
to a mechanism acting through large-scale weather effects of lunar radiation. The importance
of Jacsó's law from the point of view of long-range forecasting is stressed. Subject headings:
1. Lunar influences. 2. Weather law. 3. Precipitation periodicities. 4. Hungary.--G.T.

1. *Chlorophyll a* (Chl a) content was determined using a spectrophotometer (Shimadzu UV-160U) at 663 nm. The concentration of Chl a was calculated using the following equation: $\text{Chl a (mg/L)} = 12.7 \times \text{OD}_{663}$.

PERKES, Z.

1. The weather forecasting in Hungary has been improved recently. Its present value is 80% for half month forecasts and 50% for half month forecasts. The author finds that the existing method is not sufficient to predict the weather. The author suggests a method of forecasting the weather by using the correlation existing between the weather and the activity of the sun and cyclone activity. The author also mentions the correlation existing between the weather and the moon. Subject Headings: 1. Long range forecasting 2. Solar influences 3. Moon influences 4. Hungary.—GT

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APPROVED FOR RELEASE: 06/08/2000

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BERKES ZOLTAN

BERKES, Z

Unusual weather in April 1955, p. 187.
IDOJARAS. Budapest. Vol. 59, no. 3, May/June 1955

SOURCE: East European Accessions List (EEAL), LC, Vol. 5, No. 2,
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BERKES, Z

Szilard Zerinvary's Nap, fold, emberiseg (Sun, Earth, Mankind); a book review. p. 244

IDOJARAS. Budapest. Vol. 59, no. 4, July/Aug. 1955

SOURCE: East European Accessions List (EEAL), LC, Vol. 5, No. 2, February 1956

BERKES, Z.; ~~REDACTED~~

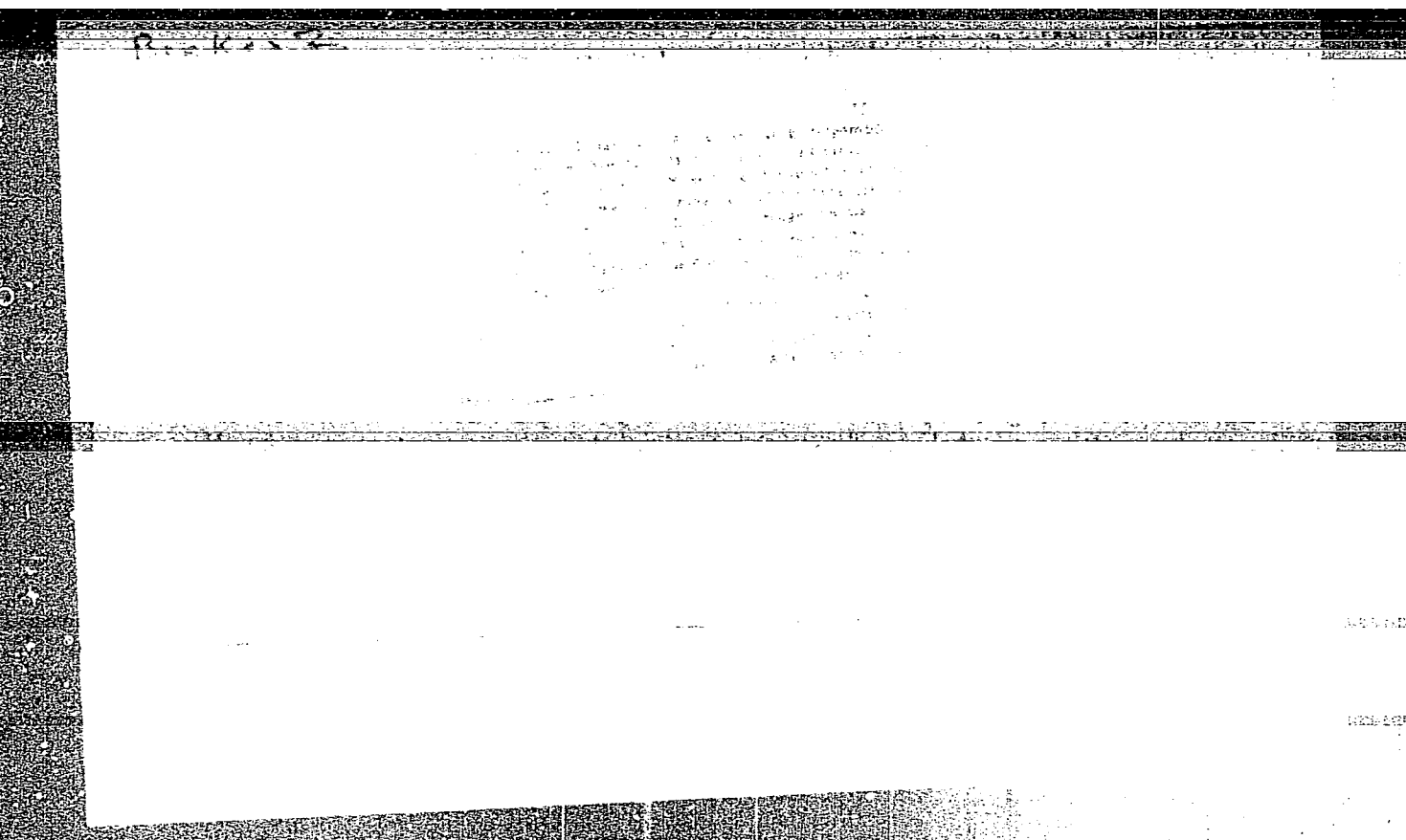
Frequency of cyclogenesis on the earth's northern hemisphere. p. 321. IDOJARAS
(Meteorologiai Intezet és Magyar Meteorologiai Társaság) Budapest. Vol. 59,
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Significant ultrapolar processes, p. 183, IDOJARAS, (Meteorologiai
Intezet es Magyar Meteorologiai Tarsasag) Budapest, Vol. 60, No. 3,
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SOURCE: East European Accessions List (EEAL) Library of Congress,
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A. DELANEY, JR.
1. The first of the two main
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